

## Iodidomesityltellurium(II) iodido-trimesitylditellurium(II)(Te—Te)

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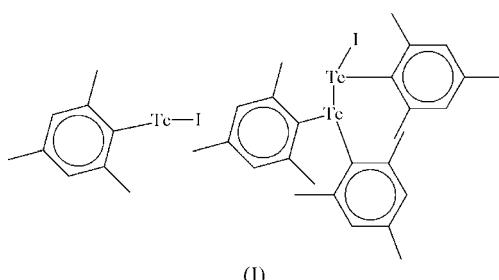
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In the title compound,  $C_9H_{11}ITe \cdot C_{27}H_{33}ITe_2$  or  $(Mes)TeI \cdot (Mes_2Te)TeI(Mes)$  ( $Mes$  is mesityl or 2,4,6-trimethylphenyl), a strong  $Te \cdots I$  interaction of  $3.3157(9)$  Å links the Te atom of an iodidomesityltellurium(II) moiety,  $(Mes)TeI$ , and an I atom of the iodidotrimesitylditellurium(II) unit,  $(Mes_2Te)TeI(Mes)$ . Further weak  $Te \cdots I$  contacts of  $4.0818(9)$  Å give rise to one-dimensional chains along the  $b$  axis in the crystal structure. An intramolecular  $C-H \cdots \pi(\text{arene})$  interaction is present in the  $(TeMes_2)TeI(Mes)$  moiety, with a  $C \cdots Cg$  distance of  $3.497(9)$  Å and a  $C-H \cdots Cg$  angle of  $142^\circ$  (where  $Cg$  is the centroid of a mesityl ring of the  $Mes_2Te$  moiety).

### Comment

Organyltellurenyl halides ( $RTeHal$ , where  $Hal$  is Cl, Br or I) are recognized as very unstable compounds which can easily undergo dismutation and disproportionation reactions. Stabilization of these species can be achieved by introducing either bulky substituents at the  $R$  moiety or additional coordinating donor groups in the system. In this way, several 2,4,6- $R'_3C_6H_2TeHal$  systems have been synthesized and structurally



characterized, as well as donor-stabilized organyltellurenyl halides (Lang *et al.*, 2006; de Oliveira *et al.*, 2006; Du Mont *et al.*, 1988; Klapötke *et al.*, 2005; Beckmann & Hesse, 2007; Beckmann *et al.*, 2007). On reacting bis(2,4,6-trimethylphenyl)ditellane ( $MesTe$ )<sub>2</sub> with one equivalent of I<sub>2</sub> in

CH<sub>2</sub>Cl<sub>2</sub>, followed by removal of the solvent *in vacuo* and recrystallization from *n*-hexane, we obtained single crystals of the title compound,  $(Mes)TeI \cdot [(Mes_2Te)TeI(Mes)]$  (where Mes is 2,4,6-trimethylphenyl), (I) (Fig. 1).

The asymmetric unit of (I) features a pseudohalide mesityltellurenyl monoiodide ( $MesTeI$ ) and a (dimesityltelluride)(iodido)(mesityl)tellurium(II)  $[(Mes)ITe(TeMes_2)]$  moiety linked by a ‘soft–soft’  $Te_3 \cdots I$ <sup>i</sup> charge-transfer interaction of  $3.3157(9)$  Å [symmetry code: (i)  $-x, -\frac{1}{2} + y, \frac{1}{2} - z$ ] (Table 1). In the  $(Mes)TeI$  fragment, the I atom is almost perpendicular to the plane containing the aromatic ring and the Te atom [ $C28-Te2-I2 = 94.5(2)^\circ$ ], as also observed in 2,4,6- $'Bu_3C_6H_2TeBr$  [93.61(5) $^\circ$ ; Beckmann & Hesse, 2007], 2,4,6- $'Bu_3C_6H_2TeCl$  [93.40(9) $^\circ$ ; Beckmann *et al.*, 2007] and 2,4,6- $'Bu_3C_6H_2TeI$  [95.75(8) $^\circ$ ; Klapötke *et al.*, 2005]. According to Ledesma *et al.* (2004), the  $(Mes)ITe(TeMes_2)$  moiety can be considered as formed by an  $(Mes)TeI$  unit stabilized by a dimesityltelluride ligand ( $Mes_2Te$ ) via a  $Te_1-Te_2$  secondary soft–soft interaction of  $3.0609(8)$  Å. The three-centre  $Te_1-Te_2-I$  system is approximately linear [171.90(3) $^\circ$ ], suggesting charge transfer from  $Mes_2Te$  to  $(Mes)TeI$ , which could explain the longer  $Te_2-I$  distance [2.9621(9) Å] compared with  $Te_3-I$  [2.7840(10) Å]. The geometry about atom Te2 can be considered as a distorted T shape, whereas the coordination geometry around atom Te1 can be described as a distorted tetrahedron with a vacant coordination site (Ledesma *et al.*, 2004).

An intramolecular  $C-H \cdots \pi(\text{arene})$  interaction is present in the  $[(Mes_2Te)TeI(Mes)]$  moiety, with a  $C7 \cdots Cg1$  distance of  $3.497(9)$  Å and a  $C7-H7A \cdots Cg1$  angle of  $142^\circ$  (where  $Cg1$  is the centroid of the C10–C15 mesityl ring) (Table 2), together with a reciprocal  $C18 \cdots Cg2$  interaction, with an interplanar  $Cg1 \cdots Cg2$  separation of  $3.39$  Å ( $Cg2$  is the centroid of the C1–C6 ring).

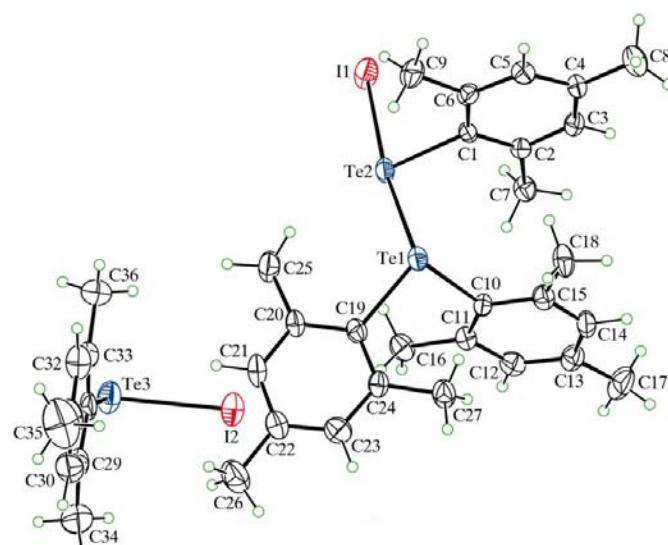
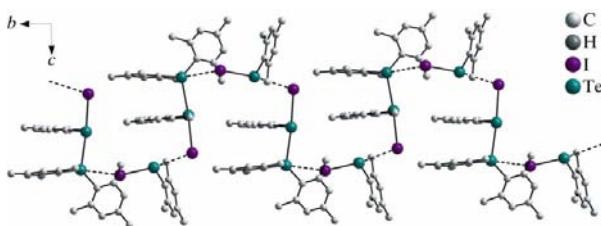


Figure 1

A view of compound (I), showing the atom-numbering scheme and 30% probability displacement ellipsoids. H atoms are shown as small spheres of arbitrary radii.

A further soft–soft weak Te<sub>1</sub>···I<sub>2</sub><sup>ii</sup> interaction of 4.0818 (9) Å [symmetry code: (ii)  $-1 + x, y, z$ ] gives rise to zigzag polymers which run along the *b* axis in the crystal structure (Fig. 2). Overall, compound (I) can be described as mesityltellurenyl monoiodide (MesTeI) units bridging the (dimesityltelluride)(iodido)(mesityl)tellurium(II) [(Mes)ITe<sub>2</sub>(TeMes<sub>2</sub>)] units in a head-to-tail fashion *via* Te···I soft–soft interactions.

**Figure 2**

A view of the one-dimensional supramolecular motif, showing the Te···I interactions (indicated as dashed lines). H atoms have been omitted for clarity.

## Experimental

Bis(2,4,6-trimethylphenyl)ditellane (MesTe)<sub>2</sub> (0.184 mmol) and iodine (0.184 mmol) were dissolved in dichloromethane (15 ml) (Du Mont *et al.*, 1988). After stirring for 30 min, the green solution was evaporated *in vacuo*. Recrystallization from *n*-hexane yielded dark-red needles of the title compound, (I).

### Crystal data

C <sub>9</sub> H <sub>11</sub> ITe·C <sub>27</sub> H <sub>33</sub> ITe <sub>2</sub>	V = 3822.3 (4) Å <sup>3</sup>
<i>M</i> <sub>r</sub> = 1113.31	Z = 4
Monoclinic, <i>P</i> <sub>2</sub> <sub>1</sub> / <i>c</i>	Mo $\text{K}\alpha$ radiation
<i>a</i> = 8.1491 (5) Å	$\mu$ = 3.91 mm <sup>-1</sup>
<i>b</i> = 15.9071 (10) Å	<i>T</i> = 297 (2) K
<i>c</i> = 29.495 (2) Å	0.28 × 0.22 × 0.19 mm
$\beta$ = 91.3790 (10) <sup>°</sup>	

### Data collection

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (*SHELXTL*; Bruker, 2001)  
 $R_{\text{int}} = 0.37$ ,  $T_{\text{max}} = 0.476$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.060$   
 $wR(F^2) = 0.127$   
 $S = 1.23$   
6720 reflections

27296 measured reflections  
6720 independent reflections  
5651 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.056$

382 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 1.24 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.57 \text{ e } \text{\AA}^{-3}$

All H atoms were placed in calculated positions using a riding model, with C—H = 0.93–0.97 Å and  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl or  $1.2U_{\text{eq}}(\text{C})$  for aryl H atoms. The methyl groups were allowed to rotate but not to tip.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT-Plus* (Bruker, 2000); program(s) used to solve structure: *SHELXTL* (Bruker, 2001); program(s) used to refine

**Table 1**  
Selected geometric parameters (Å, °).

I1—Te2	2.9621 (9)	Te1—Te2	3.0609 (8)
I2—Te3	2.7840 (10)		
C19—Te1—Te2	116.7 (2)	C1—Te2—Te1	79.9 (2)
C10—Te1—Te2	94.0 (2)	I1—Te2—Te1	171.90 (3)
C1—Te2—I1	92.3 (2)	C28—Te3—I2	94.5 (2)
C19—Te1—Te2—C1	−179.6 (3)	C33—C28—Te3—I2	88.2 (7)

**Table 2**  
Hydrogen-bond geometry (Å, °).

*Cg1* and *Cg2* are the centroids of aromatic rings C10–C15 and C1–C6, respectively.

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7A···Cg1	0.96	2.69	3.497 (9)	142
C18—H18C···Cg2	0.96	2.89	3.788 (10)	156

structure: *SHELXTL* (Bruker, 2001); molecular graphics: *ORTEP-3* (Farrugia, 1997), *DIAMOND* (Brandenburg, 2006) and *PLATON* (Spek, 2003); software used to prepare material for publication: *publCIF* (Westrip, 2007).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: GG3108). Services for accessing these data are described at the back of the journal.

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